Fate of Nitrogen During Biomass Combustion

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Introduction

- The stringent NOx emission limits for large power stations will decrease from 500mg/Nm³ to 200mg/Nm³ in 2016. New control measures will be required in order to meet these new targets.
- Understanding fuel-nitrogen behaviour during combustion is needed for the development of a firing strategy that would help achieve low NOx emissions.



Figure 1: Stages during biomass combustion: a) drying, b) devolatilisation, c) char combustion, adapted from Van Loo et al. [1].

Aims

For fuels such as pine and wheat straw:

- To study the evolution profiles of NOx precursors at different heating rates and compare the laboratory results with FG-Biomass software predictions.
- To understand how heating rate affects the split of fuel-N into volatile-N and char-N for raw fuel and torrefied fuel.

Evolution profile of NOx precursors

- During devolatilisation, the evolution profiles of NOx precursors (NH₃, HCN, HNCO) are not the same for each fuel.
- FG-BioMass software can model the evolution profiles of these NOx precursors, as shown in Figure 2.



Figure 2: Comparison of the measured and predicted NH₃ evolution profile for the devolatilisation of wheat straw at a heating rate 10°C/min.

References

[1] Diagram adapted from VAN LOO, S. & KOPPEJAN, J. 2008. The handbook of biomass combustion and co-firing, London, Earthscan

[2] Kasparbauer, R. D., 2009. The Effects of Biomass Pretreatments on the Products of Fast Pyrolisis. Graduate Theses and Dissertations, p. Paper 10064.

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Effect of heating rate on nitrogen partitioning

- Wheat straw and pine fuel were combusted using a two stage process with initial devolatilisation in a nitrogen atmosphere followed by char combustion in 12.5% oxygen in helium.
- Two heating rates were used for devolatilisation. Low heat rate (LHR) was 10°Cmin⁻¹ and high heat rate (HHR) was 1000°Cmin⁻¹.
- The nitrogen partitioning during the HHR and LHR pyrolysis were calculated by a material balance from the nitrogen content of the fuel and that of the char.





Figure 3: Changes in composition during HHR and LHR pyrolysis of wheat straw.

Figure 4: Nitrogen partitioning during the HHR and LHR pyrolysis of wheat straw.

- It was found that LHR pyrolysis produced a slightly lower proportion of volatiles than HHR pyrolysis.
- However, during LHR pyrolysis a greater proportion of the nitrogen was released in the volatiles rather than retained in the char.
- The higher proportion of nitrogen released in the volatiles is preferable to char nitrogen for several power station de-NOx strategies.





Figure 5: SEM images of pine samples (500x magnification), where a) Raw pine, b) LHR pine char, and c) HHR pine char.

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Torrefaction:



Figure 6: Breakdown of lignocellulosic components during torrefaction.

The nitrogen partitioning between the volatiles and char for the raw and torrefied pine samples that were pyrolysed at HHR was calculated by a material balance, as shown in Figure 8. Results indicated:

- from the resultant solid.

Figure 8: The perc and p		
	Pyrolysis	
Pyrolysis products	←	
Valatilaa		
Mass (g) 91.36	C _{vol} /	
N (g) 0.06	K	
0(g) 42.00		
	~	
Pine Raw Char Mass (g) 8.64	N _{char}	
N (g) 0.01	< C _{chai}	
U(g) 7.52		
	Figure 8: A co	
	-	

- volatiles and chars.
- the raw fuel.



Pyrolysis of torrefied biomass

• Improves grindability, allowing for more efficient co-firing

• Increases energy density, improving energy output compared to raw biomass and lowering transportation costs.

• Results in a hygroscopic fuel, less prone to degradation upon storage.

• Increasing the torrefaction temperature may promote the release of nitrogen

• However, higher percentage of nitrogen may be retained in the chars after torrefaction, when compared to raw biomass.

Nitrogen in char/Raw Nitrogen	%
Raw Char N/Raw N	10.79
270 Char N/Raw N	14.76
290 Char N/Raw N	13.36

centage of fuel-nitrogen retained in the chars during the pyrolysis of raw pine, pine torrefied at 270°C for 30 minutes and 290°C for 30 minutes.



torrefied pine.

Conclusions

• The combustion environment affects the fate of the nitrogen present within the fuel. Varying the heating rate directly affects the partitioning of nitrogen between

• Although torrefied fuel was found to contain less total nitrogen than the raw pine, the nitrogen concentrated within the torrefied chars was greater in comparison to