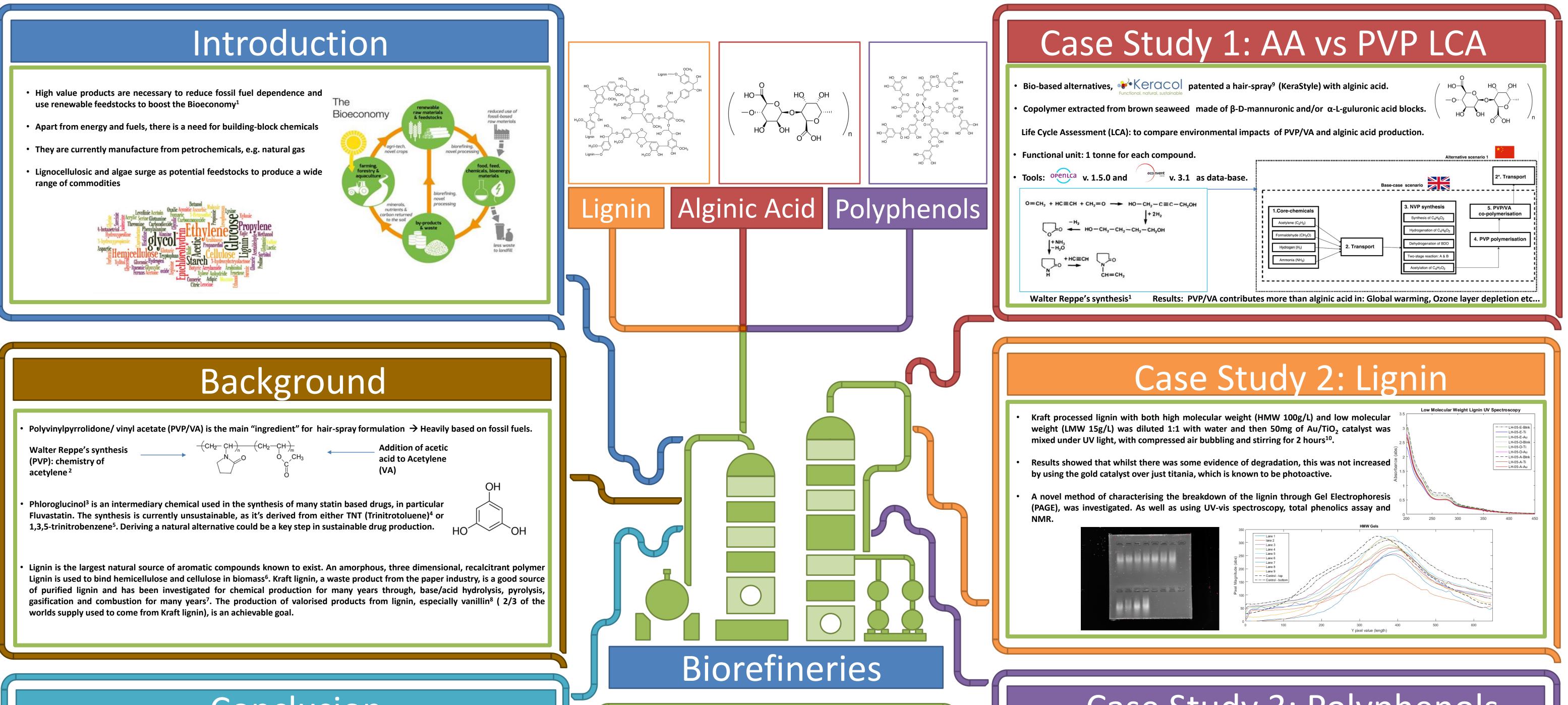


Fernando Climent Barba Calum Birch Luke Higgins

High value products from Biomass



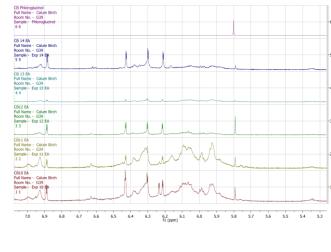
Walter Reppe's synthesis
(PVP): chemistry of
acetylene ²

Conclusion

- LCA demonstrated an elevated global warming potential due to huge dependence of natural gas in PVP/VA compared to alginic acid. This co-polymer is based on complicated industrial processes, which involves higher temperature and pressure at industrial scale. Others aspects in PVP polymerisation, such as photolysis of hydrogen peroxide by UV light or manufacturing of catalysts were not included, which will accentuate its environmental impact over alginic acid.
- Au/TiO2 photocatalyst did not show any improvement over titania catalyst on its own. The project was useful in producing a method for using PA gel electrophoresis as a qualitative method for estimating degradation of lignin. A comprehensive review of literature on the degradation of lignin was also produced.
- Synthesis of phloroglucinol from seaweed derived polyphenols is possible through a base hydrolysis reaction using sodium hydroxide. Increasing the concentration of the base produces a crude product with a high phloroglucinol presence and purity. With no other aromatics present.
- Biorefineries have the potential to develop high value products from biomass, producing complex materials and chemicals from biological sources. This study indicates the feasibility of creating these sustainable alternatives, that not only do they reduce the carbon footprint from synthesis, but further shorten complex reaction procedures, having a twinned benefit to the manufacturer.

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Case Study 3: Polyphenols

Seaweed derived Polyphenols were degraded through a base hydrolysis reaction to yield a crude product containing a naturally derived Phloroglucinol¹¹. Several experiments were ran to determine factors effecting the presence of phloroglucinol and purity of the sample. With an overall outcome indicating a 4M NaOH concentration having the most advantageous impact.

All samples were analysed using 1H NMR (500MHz) to determine the aromatic hydrogen environments within the sample. A further impactful result came from leaving the reaction stirring under Nitrogen after completion of the reflux period, but before extraction using ethyl acetate. This pre-extraction period proved effective at reducing the overall phenolic content¹².

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C019 9A - C019 9A - C019 Brch
CB18 5A - Calum Birch Rom No C39 Sample - Cp 18 FA - 22
CB16 EA Full Name - Calum Birch Nam No G99 Sample - Op 16 EA 11